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Review. multiple choice  
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pressure (also  
atmospheric) SI unit of  
force. definition & use  
of a barometer.  
standard temperature  
& pressure (STP)  
Definition of Dalton's  
law of partial  
pressures. Definitions  
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CHAPTER 11 REVIEW

Gases Class SHORT

ANSWER Answer the following questions in the space provided.

Increasing rate of effusion. List the following gases in order (Assume all gases are at the same temperature and pressure.) (c) HCl (d) 2. Explain your reasoning for the order of gases

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you chose in item 1  
above. Refer to the  
kinetic-

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This might also be a  
good time to read the  
Chapter Objectives,  
which precede the  
Review Questions.  
Chemists try to “see”  
the structure of matter  
even more closely than  
can be seen in any  
photograph. ...  
negative charge.

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Chapter 11 Modern  
Atomic Theory. 3 – .  
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atomic theory - An  
Introduction to  
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Atomic Theory Review  
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as Standing Waves  
Waveforms for  
Hydrogen Atoms  
Particle Interpretation  
of the Wave Character  
of the Electron Other  
Important Waveforms  
Overall Organization of  
Principal

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students decide if  
given statements are  
true or false. Students  
relate information  
learned about  
introductory knowledge  
gained in chemistry to  
accurately answer the  
given questions.



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and the Concept of pH

473 Determining pH

and Titrations 487

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CHAPTER 11 REVIEW

Gases SECTION 2

SHORT ANSWER

Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the

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following  
circumstances:

increase a.

temperature increases,  
volume stays the same

decrease b. volume

increases, temperature  
stays the same

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